

# Structure of the Retinal Chromophore in the hR<sub>L</sub> Intermediate of Halorhodopsin from Resonance Raman Spectroscopy<sup>†</sup>

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Received March 30, 1987; Revised Manuscript Received June 12, 1987

**ABSTRACT:** Time-resolved resonance Raman spectra of the hR<sub>L</sub> intermediate of halorhodopsin have been obtained. The structurally sensitive fingerprint region of the hR<sub>L</sub> spectrum is very similar to that of bacteriorhodopsin's L<sub>550</sub> intermediate, which is known to have a 13-cis configuration. This indicates that hR<sub>L</sub> contains a 13-cis chromophore and that an all-trans → 13-cis isomerization occurs in the halorhodopsin photocycle. hR<sub>L</sub> exhibits a Schiff base stretching mode at 1644 cm<sup>-1</sup>, which shifts to 1620 cm<sup>-1</sup> in D<sub>2</sub>O. This demonstrates that the Schiff base linkage to the protein is protonated. The insensitivity of the C-C stretching mode frequencies to N-deuteration suggests that the Schiff base configuration is anti. The 24 cm<sup>-1</sup> shift of the Schiff base mode in D<sub>2</sub>O indicates that the Schiff base proton in hR<sub>L</sub> has a stronger hydrogen-bonding interaction with the protein than does hR<sub>578</sub>.

**H**alorhodopsin is one of several photoreactive retinal-containing membrane proteins found in the halophilic bacterium *Halobacterium halobium*. These light-harvesting proteins, bacteriorhodopsin (bR),<sup>1</sup> halorhodopsin (hR), and two sensory rhodopsins, undergo cyclic photoreactions effecting transmembrane proton translocation, chloride transport, and phototaxis, respectively (Stoeckenius & Bogomolni, 1982; Lanyi, 1986; Bogomolni & Spudich, 1982; Takahashi et al., 1985; Wolff et al., 1986; Scherrer et al., 1987).

The photocycle of hR has been studied by using flash absorption techniques, and complex salt-dependent pathways have been described (Weber & Bogomolni, 1981; Tsuda et al., 1982; Schobert et al., 1983, 1986; Steiner et al., 1984; Hazemoto et al., 1984; Oesterhelt et al., 1985; Lanyi & Vodyanoy, 1986). Similar to the well-characterized photocycle of bR, the halide-dependent photocycle of hR<sub>578</sub> begins with a red-shifted intermediate (hR<sub>610</sub>) which decays into a green-absorbing intermediate which we will denote as hR<sub>L</sub>. After this point, the photocycles differ considerably, as no M-like intermediate is observed in the hR chloride transport photocycle. Additional intermediates in the latter part of the hR photocycle have been proposed and may include a red-absorbing hR<sub>640</sub> and a blue-shifted hR<sub>565</sub> (Oesterhelt et al., 1985; Lanyi & Vodyanoy, 1986).

A central problem is to determine the structural changes in the retinal chromophore and the surrounding protein that drive chloride pumping in halorhodopsin. Recently, the primary structure of halorhodopsin has been determined (Blank & Oesterhelt, 1987). Chromophore extraction studies and absorption difference spectra have suggested that an all-trans to 13-cis isomerization occurs during the photocycle of hR (Ogurusu et al., 1981; Lanyi, 1984). However, direct evidence on the in situ structure of the retinal chromophore in the photointermediates of hR has been lacking.

Resonance Raman spectroscopy is a very effective technique for studying chromophore structure in retinal pigments. The vibrational modes are sensitive to changes in molecular ge-

ometry, and the relationship between the normal mode pattern and chromophore structure has been elucidated (Smith et al., 1985, 1987a,b). Previously reported resonance Raman data have shown that the structure of the chromophore in hR<sub>578</sub> is very similar to the all-trans chromophore in bR<sub>568</sub> (Smith et al., 1984a; Maeda et al., 1985; Alshuth et al., 1985). In this paper, we present resonance Raman spectra of the long-lived hR<sub>L</sub> intermediate of halorhodopsin. The spectrum of hR<sub>L</sub> is strikingly similar to that of bacteriorhodopsin's 13-cis L<sub>550</sub> intermediate, suggesting that they have similar structures. This argues that an all-trans → 13-cis isomerization has occurred during the phototransformation of hR<sub>578</sub> to hR<sub>L</sub>.

## EXPERIMENTAL PROCEDURES

Halorhodopsin was isolated according to previously published procedures (Taylor et al., 1983). Briefly, isolated membranes from *H. halobium* strain JW-12 (containing hR but not bR) were solubilized in octyl glucoside and subjected to hydroxylapatite chromatography (Bio-Rad DNA grade) followed by chromatography over octyl-Sepharose C1-4B (Pharmacia). The pooled fractions were extensively dialyzed (48 h) to remove octyl glucoside. Samples with  $A_{280}/A_{578} \leq 2.2$  were used for Raman spectroscopy.

The hR samples were contained in a closed-loop recirculating flow system consisting of silicon tubing attached to a 1.5-mm-diameter capillary. The samples were pumped with a Cole-Parmer Model 7553-20 peristaltic pump, yielding flow rates from 40 to 500 cm/s. The optical density was kept at approximately 1.5 ODU/cm, and typical sample volumes were 1–10 mL. The temperature was maintained between 15 and 20 °C by using a stream of cold nitrogen directed on the sample reservoir.

The hR<sub>L</sub> and L<sub>550</sub> data were obtained by using a Kr<sup>+</sup> pump laser at 568.2 nm to initiate the photocycle and an Ar<sup>+</sup> probe beam at 514.5 or 501.7 nm which was spatially displaced downstream from the pump. The amount of photointermediate produced depends upon the photoalteration parameter,  $F = (3.824 \times 10^{-21})P\epsilon\phi/vd$  (Mathies et al., 1976), where  $P$  is the laser power (photons/s),  $\epsilon$  is the extinction coefficient (M<sup>-1</sup> cm<sup>-1</sup>),  $\phi$  is the quantum yield for the photochemical reaction,

<sup>†</sup> This work was supported by NIH Program Project Grant GM 27057. S.P.A.F. is an NIH postdoctoral fellow (GM 11266).

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<sup>1</sup> Abbreviations: HEPES, 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid; bR, bacteriorhodopsin; hR, halorhodopsin.

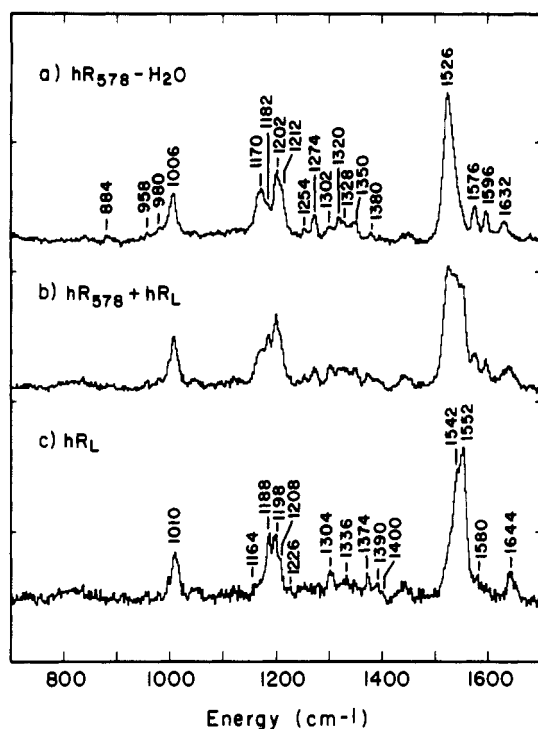


FIGURE 1: Resonance Raman spectra of halorhodopsin in 3 M NaCl/50 mM HEPES, pH 7.3. (a)  $hR_{578}$  spectrum excited at 514.5 nm; (b) pump-probe spectrum of  $hR_{578}$  +  $hR_L$  (568-nm pump and 514.5-nm probe); (c) pure  $hR_L$  spectrum obtained by subtracting (a) from (b).

$v$  is the flow velocity (cm/s), and  $d$  is the focused beam diameter (cm). To obtain  $hR_L$  spectra, an  $\sim 15$  mW 568.2-nm pump beam was cylindrically focused on the 0.15-cm-diameter capillary, giving a photoalteration parameter of  $\sim 1.0$  (flow rate 40 cm/s). The low photoalteration probe beam ( $< 2$  mW at 514.5 nm) was also cylindrically focused to give  $F \leq 0.1$ . The greatest concentration of  $hR_L$  was produced with a pump-probe spatial separation of  $\sim 0.18$  mm, corresponding to a time delay of  $\sim 450$   $\mu$ s. For  $L_{550}$  spectra, the flow rate was increased to  $\sim 400$  cm/s, and a time delay of  $\sim 25$   $\mu$ s was used. The cylindrically focused pump (80 mW, 568 nm) and probe (5 mW, 514.5 nm) beams had photoalteration values of  $\sim 0.8$  and  $< 0.1$ , respectively. The possible presence of unwanted photolysis products of  $hR_L$  was tested by lowering the pump or probe powers. No significant changes in the ethylenic band frequencies or relative intensities were observed. The absence of changes in the absorption spectrum even after prolonged irradiation showed that no significant sample degradation had occurred. Spectra of unphotolyzed  $hR_{578}$  and  $bR_{568}$  were obtained with the same apparatus by blocking the pump beam.

Raman data were acquired with a Spex 1401 double monochromator with photon counting detection interfaced to a PDP 11/23 computer. The monochromator step size was 2  $\text{cm}^{-1}$  with a 2-s dwell time, and the spectral band-pass was 5  $\text{cm}^{-1}$ . A typical spectrum is an average of 5–15 scans with no smoothing. Band deconvolution of the ethylenic frequencies was used to determine the relative concentration of  $hR_L$  at various time delays. The deconvolution gave best fits when three ethylenic bands were used at 1526, 1542, and 1552  $\text{cm}^{-1}$  with band widths constrained to 13.5  $\text{cm}^{-1}$ .

## RESULTS

Resonance Raman spectra of halorhodopsin in  $\text{H}_2\text{O}$  are presented in Figure 1. Under low photoalteration conditions, the spectrum of all-trans  $hR_{578}$  is obtained (Figure 1a) as has

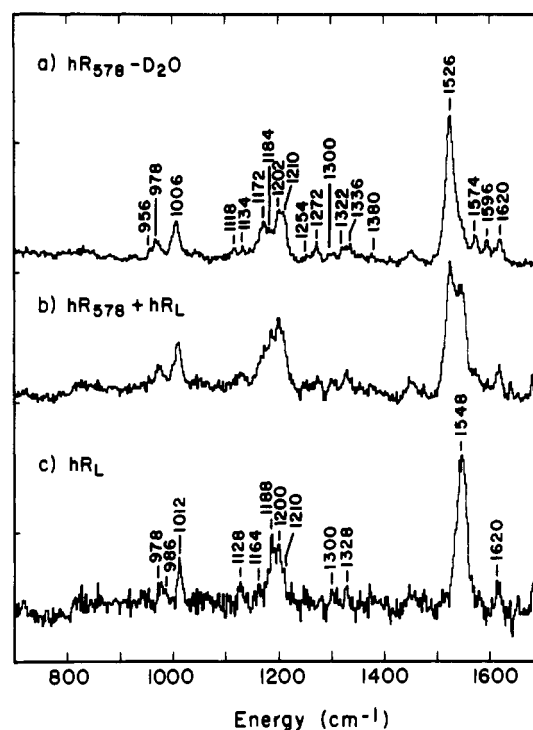


FIGURE 2: Resonance Raman spectra of deuteriated halorhodopsin (3 M NaCl/50 mM HEPES, pD 7.3). (a)  $hR_{578}$ ; (b) pump-probe spectrum of  $hR_{578}$  +  $hR_L$ ; (c) pure  $hR_L$  spectrum resulting from subtraction of (a) from (b).

been shown previously (Smith et al., 1984a; Maeda et al., 1985; Alshuth et al., 1985). Upon addition of a pump beam, a composite spectrum of  $hR_{578}$  and  $hR_L$  is produced as illustrated in Figure 1b. The 450- $\mu$ s time delay chosen for these spectra gave the largest amplitude of the  $hR_L$  ethylenic stretches. Subtraction of spectrum a from spectrum b yields the pure  $hR_L$  spectrum in Figure 1c. The subtraction parameter was determined by examining the intensities in the 1500–1526  $\text{cm}^{-1}$  ethylenic region. A correctly subtracted spectrum was obtained when no negative or positive peaks appeared in this region as would result from over- or undersubtraction of the  $hR_{578}$  ethylenic line at 1526  $\text{cm}^{-1}$ . The corresponding data for  $hR_L$  in  $\text{D}_2\text{O}$  are given in Figure 2.

To ensure that we are observing spectra of  $hR_L$ , experiments were performed with variable time delays between the pump and probe lasers ranging from 0.45 to 40 ms. The relative intensity of the sum of the 1542 and 1552  $\text{cm}^{-1}$  integrated peak areas was plotted vs time (Figure 3). These data were best fit to biexponential decay kinetics and gave  $t_{1/2}$  values of  $\sim 0.4$  and  $\sim 21$  ms. These kinetic values are in good agreement with decay constants of 0.45 and 18 ms obtained from transient absorption studies on identical preparations (Xie et al., 1986).

The  $hR_L$  spectrum is compared with that of bacteriorhodopsin's  $L_{550}$  intermediate in Figure 4. The most intense feature in the  $hR_L$  spectrum is the ethylenic doublet at 1542/1552  $\text{cm}^{-1}$ , which is similar to the ethylenic doublet of  $L_{550}$  (Figure 4b). The ethylenic doublet may originate either from the presence of two similar L species or from a pair of intense C=C stretching modes in a single L form (Alshuth & Stockburger, 1986; Argade & Rothschild, 1983). The signal-to-noise ratio in our data was insufficient to determine whether the two ethylenic components in  $hR_L$  decay independently or in parallel.

A linear correlation exists between the absorption maximum of retinal compounds and the ethylenic stretching frequencies (Rimai et al., 1973; Aton et al., 1977). For  $hR_L$ , the 1552 and 1542  $\text{cm}^{-1}$  ethylenic bands correlate with an absorption

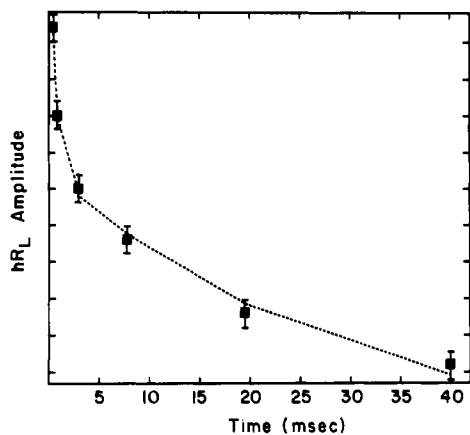


FIGURE 3: Plot of  $hR_L$  concentration vs time. The concentration of  $hR_L$  was determined from deconvoluted ethylenic band areas as described in the text. The data were fit to a biexponential decay expression yielding  $t_{1/2}$  values of  $\sim 0.4$  and  $\sim 21$  ms and respective amplitudes of 4.6 and 3.7.

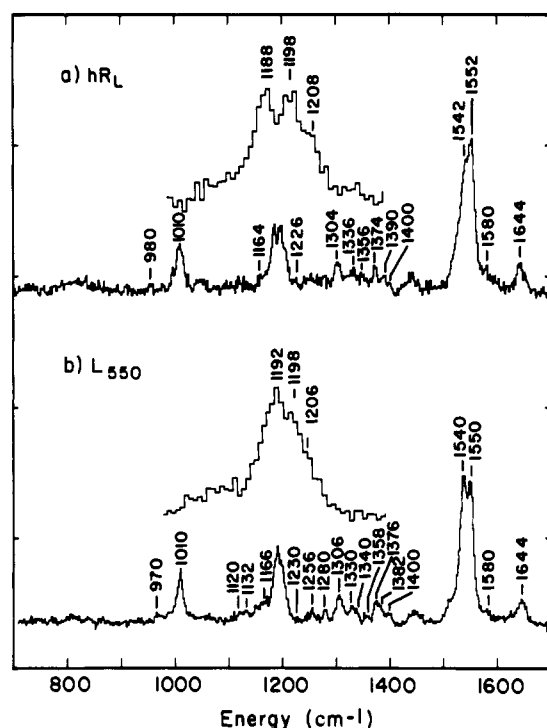


FIGURE 4: Comparison of the resonance Raman spectrum of  $hR_L$  (a) with that of bacteriorhodopsin's  $L_{550}$  intermediate (b).

maximum of  $\sim 495$  and  $\sim 540$  nm, respectively, agreeing satisfactorily with published reports of absorption maxima for  $hR_L$  ranging between 500 and 520 nm. This result, in conjunction with the Raman decay experiments of Figure 3, argues that we have obtained spectra of the same  $hR_L$  species seen in previous transient absorption experiments.<sup>2</sup>

The Schiff base stretching mode in  $hR_L$  ( $1644\text{ cm}^{-1}$ ) has the same frequency as the Schiff base mode in  $L_{550}$ . A characteristic downshift of the  $\text{C}=\text{N}$  stretch in  $\text{D}_2\text{O}$  is diagnostic for a protonated Schiff base. The  $1644\text{ cm}^{-1}$   $\text{C}=\text{N}$  stretch in  $hR_L$  (Figure 1c) shifts to  $1620\text{ cm}^{-1}$  in  $\text{D}_2\text{O}$  (Figure

2c), providing strong evidence for a protonated Schiff base linkage to the protein in  $hR_L$ . This shift is identical with the  $24\text{ cm}^{-1}$  deuterium shift observed in the Schiff base mode of  $L_{550}$  (Alshuth & Stockburger, 1986).

In retinals and retinal-containing proteins, the  $\text{C}-\text{C}$  skeletal stretching modes in the  $1100\text{--}1300\text{ cm}^{-1}$  fingerprint region provide sensitive markers for chromophore geometry. Characteristic frequency differences are observed between *all-trans*- and *13-cis*-retinal (Curry et al., 1982, 1984), between their protonated Schiff base derivatives in vitro (Smith et al., 1985, 1987b), and between  $bR_{568}$  and  $bR_{548}$  (Smith et al., 1987a,b). In  $L_{550}$ , the  $\text{C}_{14}-\text{C}_{15}$  stretching mode has been assigned at  $1155\text{ cm}^{-1}$  in the infrared (Gerwert & Siebert, 1986) and at  $1172\text{ cm}^{-1}$  in the Raman spectrum (Smith et al., 1986). In the present work, a broad, weak band is observed with a maximum at  $\sim 1166\text{ cm}^{-1}$ , which we attribute to the  $\text{C}_{14}-\text{C}_{15}$  mode following the earlier assignments. The  $1192\text{ cm}^{-1}$  mode is assigned as the " $\text{C}_{10}-\text{C}_{11}$  stretch", and the shoulder at  $1198\text{ cm}^{-1}$  is the  $\text{C}_8-\text{C}_9$  mode (Smith, 1985). Compared to *all-trans*  $bR_{568}$ , the characteristic feature of the  $L_{550}$  spectrum is the reduced frequency of the  $\text{C}_{14}-\text{C}_{15}$  mode ( $1201\text{ cm}^{-1} \rightarrow \sim 1166\text{ cm}^{-1}$ ) which pushes the  $\text{C}_{10}-\text{C}_{11}$  stretch up from  $\sim 1170\text{ cm}^{-1}$  in  $bR_{568}$  to  $1192\text{ cm}^{-1}$  in  $L_{550}$  (Smith et al., 1985). The very close correspondence between the  $\sim 1166$ ,  $1192$ , and  $1198\text{ cm}^{-1}$  modes in  $L_{550}$  with the  $1164$ ,  $1188$ , and  $1198\text{ cm}^{-1}$  modes in  $hR_L$  suggests that similar assignments apply and that  $hR_L$  is also *13-cis*.

The configuration about the  $\text{C}=\text{N}$  bond can be determined from the deuterium sensitivity of the  $\text{C}_{14}-\text{C}_{15}$  stretch in protonated Schiff bases (Smith et al., 1984b). In  $\text{C}=\text{N}$  syn compounds,  $\text{C}_{14}-\text{C}_{15}$  stretch and  $\text{N}-\text{H}$  rock are strongly coupled so that deuteration results in a  $40\text{--}60\text{ cm}^{-1}$  upshift of the  $\text{C}_{14}-\text{C}_{15}$  stretch. In  $\text{C}=\text{N}$  anti compounds, there is little coupling of the  $\text{C}_{14}-\text{C}_{15}$  stretch with the  $\text{N}-\text{H}$  rock, and therefore, deuteration has no effect on the  $\text{C}_{14}-\text{C}_{15}$  stretch frequency. The lack of a significant shift in any of the  $1100\text{--}1400\text{ cm}^{-1}$  fingerprint modes including the  $1164\text{ cm}^{-1}$  band upon deuteration suggests that  $hR_L$  has an anti  $\text{C}=\text{N}$  configuration.

## DISCUSSION

The remarkable similarity between the  $hR_L$  and  $L_{550}$  resonance Raman spectra (Figure 4) argues strongly that the chromophore in  $hR_L$  is *13-cis* and that the chromophores in both species experience nearly identical protein perturbations. The  $\text{C}-\text{C}$  stretching modes in  $hR_L$  have been assigned by comparison with previous assignments for  $L_{550}$  and  $bR_{548}$  (Smith, 1985; Smith et al., 1987b). The key diagnostics for *13-trans* to *13-cis* isomerization are the upshift of the intense  $\text{C}_{10}-\text{C}_{11}$  stretching mode (from  $\sim 1170$  to  $\sim 1190\text{ cm}^{-1}$ ) and the drop of the  $\text{C}_{14}-\text{C}_{15}$  stretch due to a decrease in  $\text{C}_{14}-\text{C}_{15}$  stretch-skeletal bend interaction in the *13-cis* geometry (Curry et al., 1984; Smith et al., 1985, 1987b). This is precisely the pattern observed for the  $\text{C}-\text{C}$  modes in  $hR_L$ . The frequency of the  $\text{C}_{14}-\text{C}_{15}$  stretch can also be used to examine the conformation about this bond. Previous calculations have shown that the  $\text{C}_{14}-\text{C}_{15}$  stretch in *s-cis* conformers experiences a dramatic ( $\sim 70\text{ cm}^{-1}$ ) downshift as compared to the  $\text{C}_{14}-\text{C}_{15}$  *s-trans* isomer. The  $\sim 1164\text{ cm}^{-1}$  value for the  $\text{C}_{14}-\text{C}_{15}$  stretch in  $hR_L$  is consistent with a  $\text{C}_{14}-\text{C}_{15}$  *s-trans* molecule. This result does not support a recently proposed model for chloride transport in halorhodopsin invoking a *14-s-cis* structure (Oosterhelt et al., 1986).

In light of the correspondences between the  $hR_L$  and  $L_{550}$  spectra, it is of interest to recall the similarities and differences between  $hR_{578}$  and  $bR_{568}$ . Both chromophores are *all-trans*-

<sup>2</sup> Several studies have suggested the existence of an  $hR_{520} \rightleftharpoons hR_{640}$  equilibrium in the salt-dependent hR photocycle (Oosterhelt et al., 1985; Lanyi & Vodyanov, 1986). No evidence for this equilibrium was found in the present study as judged by the lack of an ethylenic frequency corresponding to a  $640\text{-nm}$  form. In addition, transient absorption work on our samples has shown no evidence for a red-shifted intermediate in the latter part of the hR photocycle (Xie et al., 1986).

protonated Schiff bases; however, the Schiff base stretching frequency in hR<sub>578</sub> (1632 cm<sup>-1</sup>) is considerably lower than the corresponding frequency in bR<sub>568</sub> (1641 cm<sup>-1</sup>) (Smith et al., 1987a), and the resulting deuterium shift is less (12 cm<sup>-1</sup> vs 18 cm<sup>-1</sup>). This decrease in C=N stretch/N-H rock coupling has been attributed to a decreased hydrogen-bonding interaction between the Schiff base proton and its anion in hR<sub>578</sub> (Smith et al., 1984a; Alshuth et al., 1985; Maeda et al., 1985). Also, there is an increase in the C=N stretching frequency of hR<sub>578</sub> upon substitution of nitrate for chloride (Maeda et al., 1985; S.P.A. Fodor and R. A. Mathies, unpublished results). This suggests that the reduced hydrogen-bonding interaction in hR<sub>578</sub> is caused by some chloride-dependent alteration of the Schiff base environment. Recent work on the mechanism of the opsin shift in bacteriorhodopsin has suggested that a large part of the opsin shift (~3500 cm<sup>-1</sup>) is due to a decrease in the hydrogen-bonding interaction between the Schiff base proton and an anion in the protein environment (Lugtenburg et al., 1986; Spudich et al., 1986; Baasov & Sheves, 1986). The decreased deuterium effect on the Schiff base mode of hR<sub>578</sub> suggests that the extra ~300 cm<sup>-1</sup> red shift for hR (as compared to bR) may be due to a further decrease in the hydrogen-bonding interaction. This provides strong evidence that the mechanism of the opsin shift in hR is also due primarily to changes in Schiff base hydrogen bonding.

When hR<sub>578</sub> is converted to hR<sub>L</sub>, the Schiff base mode shifts from 1632 to 1644 cm<sup>-1</sup>, and we observe a large 24 cm<sup>-1</sup> downshift of the C=N stretch upon deuteration. Thus, the hydrogen-bonding environment has gone from a relatively weak one in hR<sub>578</sub> to a stronger one in hR<sub>L</sub>. Furthermore, the D<sub>2</sub>O-induced shift in hR<sub>L</sub> is the same as that observed in L<sub>550</sub>, suggesting that the perturbation responsible for the decrease in hydrogen bonding in hR<sub>578</sub> has been removed in hR<sub>L</sub>. If the weak hydrogen bonding of the Schiff base in hR<sub>578</sub> is due to some direct or indirect interaction with chloride, then the stronger hydrogen bond in hR<sub>L</sub> suggests that this interaction has been lost.

#### ACKNOWLEDGMENTS

We thank W. Stoeckenius for helpful discussions and Kathleen McGinnis and Martha Marvin for preparing the hR samples.

**Registry No.** D<sub>2</sub>, 7782-39-0.

#### REFERENCES

- Alshuth, T., & Stockburger, M. (1986) *Photochem. Photobiol.* 43, 55-66.
- Alshuth, T., Stockburger, M., Hegemann, P., & Oesterhelt, D. (1985) *FEBS Lett.* 179, 55-59.
- Argade, P. V., & Rothschild, K. J. (1983) *Biochemistry* 22, 3460-3466.
- Aton, B., Doukas, A. G., Callender, R. H., Becher, B., & Ebrey, T. G. (1977) *Biochemistry* 16, 2995-2999.
- Baasov, T., & Sheves, M. (1986) *Biochemistry* 25, 5249-5258.
- Blanck, A., & Oesterhelt, D. (1987) *EMBO J.* 6, 265-273.
- Bogomolni, R. A., & Spudich, J. L. (1982) *Proc. Natl. Acad. Sci. U.S.A.* 79, 6250-6254.
- Curry, B., Broek, A., Lugtenburg, J., & Mathies, R. A. (1982) *J. Am. Chem. Soc.* 104, 5274-5286.
- Curry, B., Palings, I., Broek, A., Pardo, J. A., Mulder, P. J., Lugtenburg, J., & Mathies, R. A. (1984) *J. Phys. Chem.* 88, 688-702.
- Gerwert, K., & Siebert, F. (1986) *EMBO J.* 5, 805-811.
- Hazemoto, N., Kamo, N., Kabatake, Y., Tsuda, M., & Terayama, Y. (1984) *Biophys. J.* 45, 1073-1077.
- Lanyi, J. K. (1984) *FEBS Lett.* 175, 337-342.
- Lanyi, J. K. (1986) *Annu. Rev. Biophys. Biophys. Chem.* 15, 11-28.
- Lanyi, J. K., & Vodyanoy, V. (1986) *Biochemistry* 25, 1465-1470.
- Lugtenburg, J., Muradin-Szweykowska, M., Heeremans, C., Pardo, J. A., Harbison, G. S., Herzfeld, J., Griffin, R. G., Smith, S. O., & Mathies, R. A. (1986) *J. Am. Chem. Soc.* 108, 3104-3105.
- Maeda, A., Ogurusu, T., Yoshizawa, T., & Kitagawa, T. (1985) *Biochemistry* 24, 2517-2521.
- Mathies, R., Oseroff, A. R., & Stryer, L. (1976) *Proc. Natl. Acad. Sci. U.S.A.* 73, 1-5.
- Oesterhelt, D., Hegemann, P., & Tittor, J. (1985) *EMBO J.* 4, 2351-2356.
- Oesterhelt, D., Hegemann, P., Tavan, P., & Schulten, K. (1986) *Eur. Biophys. J.* 14, 123-129.
- Ogurusu, T., Maeda, A., Sasaki, N., & Yoshizawa, T. (1981) *J. Biochem. (Tokyo)* 90, 1267-1273.
- Rimai, L., Heyde, M. E., & Gill, D. (1973) *J. Am. Chem. Soc.* 95, 4493-4501.
- Scherrer, P., McGinnis, K., & Bogomolni, R. A. (1987) *Proc. Natl. Acad. Sci. U.S.A.* 84, 402-406.
- Schobert, B., Lanyi, J. K., & Cragoe, E. J., Jr. (1983) *J. Biol. Chem.* 258, 15158-15164.
- Schobert, B., Lanyi, J. K., & Oesterhelt, D. (1986) *J. Biol. Chem.* 261, 2690-2696.
- Smith, S. O. (1985) Ph.D. Dissertation, University of California, Berkeley, CA.
- Smith, S. O., Marvin, M. J., Bogomolni, R. A., & Mathies, R. A. (1984a) *J. Biol. Chem.* 259, 12326-12329.
- Smith, S. O., Myers, A. B., Pardo, J. A., Winkel, C., Mulder, P. P. J., Lugtenburg, J., & Mathies, R. (1984b) *Proc. Natl. Acad. Sci. U.S.A.* 81, 2055-2059.
- Smith, S. O., Lugtenburg, J., & Mathies, R. A. (1985) *J. Membr. Biol.* 85, 95-110.
- Smith, S. O., Hornung, I., van der Steen, R., Pardo, J. A., Braiman, M. S., Lugtenburg, J., & Mathies, R. A. (1986) *Proc. Natl. Acad. Sci. U.S.A.* 83, 967-971.
- Smith, S. O., Braiman, M. S., Myers, A. B., Pardo, J. A., Courtin, J., Winkel, C., Lugtenburg, J., & Mathies, R. A. (1987a) *J. Am. Chem. Soc.* 109, 3108-3125.
- Smith, S. O., Pardo, J. A., Lugtenburg, J., Mathies, R. A. (1987b) *J. Phys. Chem.* 91, 804-819.
- Spudich, J. L., McCain, D. A., Nakanishi, K., Okabe, M., Shimizu, N., Rodman, H., Honig, B., & Bogomolni, R. A. (1986) *Biophys. J.* 49, 479-483.
- Steiner, M., Oesterhelt, D., Arikawa, M., & Lanyi, J. K. (1984) *J. Biol. Chem.* 259, 2179-2184.
- Stoeckenius, W., & Bogomolni, R. A. (1982) *Annu. Rev. Biochem.* 51, 587-616.
- Takahashi, T., Tomioka, H., Kamo, N., & Kobatake, Y. (1985) *FEMS Microbiol. Lett.* 28, 161-164.
- Taylor, M. E., Bogomolni, R. A., Weber, H. J. (1983) *Proc. Natl. Acad. Sci. U.S.A.* 80, 6172-6176.
- Tsuda, M., Hazemoto, N., Kondo, M., Kamo, N., Kobatake, Y., & Terayama, Y. (1982) *Biochem. Biophys. Res. Commun.* 108, 970-976.
- Weber, H. J., Bogomolni, R. A. (1981) *Photochem. Photobiol.* 33, 601-608.
- Wolff, E. K., Bogomolni, R. A., Scherrer, P., Hess, B., & Stoeckenius, W. (1986) *Proc. Natl. Acad. Sci. U.S.A.* 83, 7272-7276.
- Xie, A., Lozier, R. H., & Bogomolni, R. A. (1986) *Biophys. J.* 49, 209a.